Photoelectrochemical (PEC) Water Splitting

Synthesis of Mn Doped CdS Nanostructures Nimrah Tahir¹, Manzar Sohail¹, Naila Jabeen²

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Introduction:

Photoelectrochemical (PEC) water splitting by sunlight irradiation has gained much attention for generation of renewable hydrogen. Controlling the properties of semiconducting materials for PEC water splitting is the main concern to achieve higher efficiency of solar light conversion.¹ Among many metal sulfides cadmium sulfide belongs to II-VI compound family is considered for number of



studies for the development of CdS solar cells, due to its high absorption co efficient, stability and low cost.²

Experimental Set Up



30 60 2.0 (°) 2 0(°)

Figure 1:XRD graph a)diff temp b)diffpH c)diff SDS amount d) diff concentrations





PEC water splitting Results



Characterization

The diffraction peaks indexed as (002) and (110) at 28.18° and 43.91° respectively confirmed the formation of CdS There is no extra peak of manganese sulfide that is the evidence of Mn doping.³

Crystalline structure was also evident from high resolution TEM. 10-20 nm nanostructures of Mn doped CdS with high resolution fringes are evident from Fig 3.



Figure 2: SEM images of Mn doped CdS (a) Nano cubes (b) Nano worms (c) nano spheres (d) Nano petals (e) Nano balls (f) Nano Plates







Fig 4: a) LSV b) CA graphs c) PEC stability

Conclusions:

	44.11	110	6.94
Temp (80)	28.15	002	9.51
	43.91	110	5.97
рН 8	28.06	002	5.49
		110	
10 % doping	28.06	002	4.37
	44.11	110	6.94





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